Metallomics

PAPER

Cite this: *Metallomics,* 2018, 10, 854

Received 9th March 2018, Accepted 6th June 2018

DOI: 10.1039/c8mt00052b

rsc.li/metallomics

Systems impact of zinc chelation by the epipolythiodioxopiperazine dithiol gliotoxin in Aspergillus fumigatus: a new direction in natural product functionality†

Aliabbas A. Saleh, \mathbf{D}^a Gary W. Jones, \mathbf{D}^{ab} Frances C. Tinley,^a Stephen F. Delaney,^a Sahar H. Alabbadi, \mathbb{D}^a Keith Fenlon,^a Sean Doyle \mathbb{D}^{*a} and Rebecca A. Owens \mathbb{D}^{*a}

The non-ribosomal peptide gliotoxin, which autoinduces its own biosynthesis, has potent anti-fungal activity, especially in the combined absence of the gliotoxin oxidoreductase GliT and bis-thiomethyltransferase GtmA. Dithiol gliotoxin (DTG) is a substrate for both of these enzymes. Herein we demonstrate that DTG chelates Zn^{2+} (m/z 424.94), rapidly chelates Zn^{2+} from $Zn(4-(2-pyridylazo)$ -resorcinol) ($Zn(PAR)_{2}$) and also inhibits a Zn^{2+} -dependent alkaline phosphatase (AP). Zn^{2+} addition rescues AP function following DTG-associated inhibition, and pre-incubation of DTG with Zn^{2+} completely protects AP activity. Zn^{2+} $(1-50 \mu)$ also significantly relieves the potent gliotoxin-mediated inhibition of Aspergillus fumigatus Δ gliT:: Δ gtmA (p < 0.05), which infers in vivo dithiol gliotoxin-mediated sequestration of free Zn²⁺ or chelation from intracellular metalloenzymes as inhibitory mechanisms. Quantitative proteomic analysis revealed that excess Zn^{2+} alters the effect of gliotoxin on A. fumigatus ΔqlT , with differential abundance of secondary metabolism-associated proteins in the combinatorial condition. GtmA abundance increased 18.8 fold upon co-addition of gliotoxin and Zn^{2+} compared to gliotoxin alone, possibly to compensate for disruption to GtmA activity, as seen in in vitro assays. Furthermore, DTG effected significant in vitro aggregation of a number of protein classes, including Zn^{2+} -dependent enzymes, while proteins were protected from aggregation by pre-incubating DTG with Zn^{2+} . We conclude that DTG can act in vivo as a Zn^{2+} chelator, which can significantly impede A. fumigatus growth in the absence of GliT and GtmA. **PAPER**
 Systems impact of zinc chelation by the
 epipolythiodioxopiperazine dithiol gliotoxin in
 epipolythiodioxopiperazine dithiol gliotoxin in
 Aspergillus furmigatus: a new direction in natural
 Product func

Significance to metallomics

Dithiol gliotoxin is a near-terminal biosynthetic intermediate from the gliotoxin biosynthetic pathway in the human pathogen Aspergillus fumigatus. Chemically reduced gliotoxin, dithiol gliotoxin (DTG), is revealed as a biological zinc chelator, and conversely, zinc can relieve the hitherto cryptic fungal autotoxicity of DTG. There is a systems-wide impact of zinc chelation by DTG on the fungal proteome, and we suggest it is DTG, as opposed to gliotoxin, which chelates zinc from metalloproteins. Since gliotoxin can be sequestered by both fungi and bacteria, our findings infer a new avenue to interfere with, and exploit, cellular zinc homeostasis in microorganisms.

Introduction

Gliotoxin and holomycin are microbial natural products, which are produced by fungal and bacterial spp., respectively $(Fig, 1)$.^{1–3} Both are low molecular mass metabolites, and each contains a

disulphide bridge formed by the action of oxidoreductases, namely GliT and HlmI, on the respective dithiol precursor (Fig. 1).^{4–6} Disulphide bridge formation is essential for microbial self-protection against these reactive dithiol intermediates, and is a pre-requisite for the Major Facilitator Superfamily transporter GliA-mediated secretion of gliotoxin by Aspergillus fumigatus.⁵⁻⁸ Both metabolites are also present as bis-thiomethylated forms, and gliotoxin bis-thiomethyltransferase GtmA converts dithiol gliotoxin (DTG) to bis-dethiobis(methylthio)gliotoxin (BmGT) in A. fumigatus, whereas the origin of the cognate activity against dithiol holomycin in Streptomyces clavuligeris is unknown (Fig. 1).^{9,10} Bernardo et al. have shown that upon uptake by eukaryotic cells,

^a Department of Biology, Maynooth University, Co. Kildare, Ireland.

E-mail: sean.doyle@mu.ie, rebecca.owens@mu.ie; Tel: +353-1-708-3858, +353-1-708-3839

 b Centre for Biomedical Research, School of Clinical and Applied Sciences,</sup> Leeds-Beckett University, Leeds LS1 3HE, UK

[†] Electronic supplementary information (ESI) available. See DOI: 10.1039/c8mt00052b

Fig. 1 Unrelated microbial dithiol-containing compounds undergo equivalent biotransformations, and appear to be co-substrates for two enzyme functionalities. (A) In Aspergillus fumigatus, DTG is either oxidised to gliotoxin via gliotoxin oxidoreductase GliT or bis-thiomethylated by gliotoxin bis-thiomethyltransferase GtmA. (B) In Streptomyces clavuligeris, dithiol holomycin undergoes oxidoreductase HlmI-mediated conversion to holomycin. The enzyme which catalyses S,S'-dimethyl holomycin formation remains to be identified.

gliotoxin is chemically reduced to the dithiol form by intracellular glutathione $(GSH)^{11}$ Carberry et al. revealed significantly elevated intracellular GSH in gliotoxin-sensitive A. fumigatus $\Delta g l i T$ and that Saccharomyces cerevisiae Δ gsh1, deficient in intracellular GSH, was resistant to exogenous gliotoxin.12 The epipolythiodioxopiperazine (ETP) gliotoxin autoinduces its own biosynthesis by activating gli biosynthetic gene cluster expression, and BmGT formation has been shown to result in gli cluster attenuation in A. fumigatus, which results in the repression of gliotoxin biosynthesis.^{9,13,14} Methylation of dithiol holomycin has been proposed as a back-up plan for self-protection, and Dolan et al. revealed that A. fumigatus $\Delta gliT$: $\Delta gtmA$ is significantly more sensitive to exogenous gliotoxin than a $gliT$ -deficient mutant.¹⁵ This suggests that the combined absence of both self-protection and negative regulation of DTG biosynthesis (or its intracellular presence) results in potent growth retardation – the precise cause of which is unknown. 10,15

Gliotoxin, other ETPs and dithiolopyrrolones have been shown to inhibit the activity of many enzymes, and functionality of specific proteins.¹⁶⁻²¹ Several studies have demonstrated that the disulphide moiety of gliotoxin is responsible for most of the associated bioactivity of this molecule, whereas the S-methylated BmGT molecule is relatively inactive.^{17,22} In general, reactivity of gliotoxin towards protein thiols, damage from redox cycling, and Zn^{2+} ejection have been proposed as the respective mechanisms whereby protein functionality is altered. Interestingly, addition of the reducing agent, L-dithiothreitol (DTT), significantly augmented the inhibitory activity of gliotoxin towards farnesyltransferase, and either DTT or GSH augmented gliotoxin-mediated inhibition of equine alcohol dehydrogenase.^{16,17} Notably, neither of the aforementioned studies posited DTG-mediated Zn^{2+} chelation from either Zn^{2+} -dependent enzyme as the inhibitory mechanism.

In combination, these observations led us to hypothesise that, acting as a potent Zn^{2+} chelator, DTG (and not gliotoxin), could exhibit an equivalent mechanism of action towards Zn²⁺-dependent enzyme systems. Moreover, DTG interference with Zn^{2+} availability or Zn^{2+} -dependent enzyme activity in A. fumigatus $\Delta g l i T$: $\Delta g t m A$ could be the basis for observed and extreme growth retardation.¹⁵ However, it is also essential to consider a mechanistic reciprocity between DTG and Zn^{2+} . Consequently, while DTG may chelate the cation and impede Zn^{2+} -mediated enzyme activities, excess or available metal ion may provide protection against intracellular DTG. Indeed, zinc salts have been successfully used to reverse ovine and bovine facial eczema associated with exposure to fungal ETPs such as sporidesmin A, a related disulphide-containing metabolite secreted by Pithomyces chartarum.²³ It is postulated that Zn^{2+} chelates formed with sporidesmin A may attenuate its unwanted biological effects. Interestingly, Woodcock et al. presented mass spectrometric evidence of Zn^{2+} [sporidesmin] chelates, and noted that $\text{Zn}^{2+}[$ gliotoxin] chelates, following NaBH₄-mediated reduction, also existed, namely [2gliotoxin + Zn]²⁻, [2gliotoxin + $\text{Zn} + \text{Na}$ ⁻ and [gliotoxin + ZnCl]⁻ (m/z 427).²⁴ These authors speculated that that mono-ligands with halide coordination were the preferable complex form, however corresponding mass spectra and fragmentation patterns were not presented. Paper Were View Methods Article is the contribution spectra Creation School and Article. Distribution of the Creation Crea

The ZafA transcriptional regulator, which can induce expression of the Zn^{2+} transporters *zrfA*, *zrfB* and *zrfC*, regulates zinc homeostasis in A. fumigatus and is essential for virulence.²⁵⁻²⁸ ZrfC is the key essential transporter which effects Zn^{2+} acquisition by A. fumigatus, while ZrfA and ZrfB play accessory and nonessential roles.²⁶ Interestingly, transporter ZrfB (AFUA 2G03860) was found to be significantly increased in abundance $(\log_2$ -fold increase: 1.31) in long-term A. fumigatus Δg tmA cultures, which suggests interplay between dysregulated gliotoxin biosynthesis and Zn^{2+} homeostasis.²⁹

Herein, for the first time we reveal DTG as a Zn^{2+} chelator which can specifically and significantly inhibit Zn^{2+} -dependent metalloenzyme activity. Moreover, we demonstrate that Zn^{2+} significantly reverses the inhibitory effects of gliotoxin on A. fumigatus $\Delta gliT$: $\Delta gtmA$, which implicates intracellular Zn^{2+} chelation as a potential growth inhibitory strategy against this pathogen.¹⁵ Significant proteomic remodelling in A. fumigatus $\Delta g l i T$ in response to gliotoxin versus $\text{Zn}^{2+}/\text{gliotox}$ in exposure further illuminates a hitherto unanticipated in vivo interaction between DTG and Zn^{2+} .

Methods

High resolution mass spectrometry detection of gliotoxin complexed to Zn^{2+}

Tris(2-carboxyethyl)phosphine hydrochloride (TCEP)-reduced gliotoxin (DTG; 300 μ M final) was mixed with ZnSO₄.7H₂O in the presence of formic acid $(0.1\% (v/v))$ to achieve 0.5, 1 and 3-fold molar equivalents of Zn^{2+} to DTG. Samples were passed through $0.22 \mu m$ spin filters and directly injected onto a Thermo Q-Exactive Mass spectrometer (5 μ l min⁻¹). Assessment of DTG-Zn²⁺

Metallomics Paper

Assessment of in vitro Zn^{2+} chelation by DTG

Zinc binding assays were performed as described in Chan et al. with some modifications.³⁰ Titrations were carried out in PBS pH 7.4 (1 ml final volume). 4-(2-Pyridylazo)-resorcinol (PAR) (80 μ l; 1.5 mM) was added with 60 μ l 1 mM ZnSO₄, per ml to yield final ratio of $2:1$ (PAR: Zn^{2+}). DTG (2.5 mM in methanol) was prepared by 60 min pre-incubation with 12.5 mM TCEP in 100 µl final volume and was then added in increasing concentration to yield 1, 2, 3, 4 and 5 molar equivalents of DTG to $\text{Zn}(\text{PAR})_2$ (final concentrations: 60 μ M Zn²⁺, 120 μ M PAR, 60-300 μ M DTG). 12.5 mM TCEP in methanol and oxidised gliotoxin (2.5 mM) were used as negative controls. After addition of all components, absorbance spectra were recorded between 200–800 nm. Metaleonics

complex formation was creduced by both positive and negative

Less Articles Articles. Propose and the state of the st

Alkaline phosphatase (AP) and GtmA enzyme assays

 $\text{Zn}^{2+}\text{-Dependent AP from bovine intestinal mucosa (AP; 1.44 U ml}^{-1}$ (Sigma-Aldrich)) in PBS was pre-incubated with the following agents to determine the extent of enzyme inhibition: 5 mM EDTA, 150 μ M gliotoxin (GT), $10-150 \mu M$ DTG (TCEP-reduced GT), $50 \mu M$ TPEN (N,N,N',N'-tetrakis(2-pyridylmethyl)ethane-1,2-diamine), 50 μM-5 mM DTT, or 50 μ M-5 mM GSH for 15 min in triplicate. Control or treated AP (50 μ l per reaction) was added to 1 ml p-nitrophenol phosphate (pNPP) (5 mM; dissolved in PBS with 50 mM glycine pH 9.5) and assayed in triplicate at 37 $^{\circ}$ C for 30 min. Enzyme reactions were terminated by addition of 20 mM NaOH (5 ml). AP activity was determined by p -nitrophenol (pNP) detection at 405 nm following pNPP hydrolysis. The protective effect of preincubating DTG or TPEN with Zn^{2+} was also evaluated using this assay system. DTG (50 μ M) or TPEN (50 μ M) were pre-incubated with Zn^{2+} (25–100 µM) for 15 min prior to addition of AP enzyme and the rest of the assay was performed as before. To test the capacity of Zn^{2+} to rescue DTP-associated loss of AP activity, AP was pre-incubated with DTG (50 μ M) for 15 min, followed by addition of Zn^{2+} (0.1-1 mM) for a further 15 min, prior to addition of pNPP substrate.

GtmA activity was determined as described in Dolan et al. to evaluate the effect of Zn^{2+} (molar ratio 0.0001–10 Zn^{2+} : 1 DTG) on the formation of BmGT.¹⁵ Briefly, S-adenosylmethionine (SAM) (1 μ M final) and TCEP-reduced gliotoxin (250 μ M final) were combined, along with PBS. Zn^{2+} was added at a range of concentrations (0.025, 0.25, 2.5, 25, 125, 250 µM and 2.5 mM), representing a molar ratio ranging from 0.0001 to $10:1 Zn^{2+}$: DTG. A control was also prepared with no Zn^{2+} added. GtmA $(0.5 \mu M)$ was added to the mixture followed by incubation at 37 °C for 5 min. Reactions were stopped by protein precipitation using TCA (final 15%) and incubation on ice for 20 min. Clarified mixtures were analysed by RP-HPLC and absorbance monitored at 254 nm to determine the concentration of BmGT and monomethyl gliotoxin (MmGT).

Assessment of the effect of Zn^{2+} on alkylation of DTG

Iodoacetamide (IAA) was used as an alkylating agent to investigate the effect of Zn^{2+} on alkylation of DTG. IAA stock (50 mM) was prepared in 50 mM ammonium bicarbonate. 3 mM DTG was prepared by 60 min pre-incubation of gliotoxin with 12.5 mM TCEP. DTG (0.3 mM final) was mixed with $ZnSO₄$ at a range of concentrations (0.1, 0.15, 0.3, 0.6, 0.9 mM) to achieve molar equivalents of Zn^{2+} to DTG of 0.33, 0.5, 1, 2 and 3, respectively. IAA was subsequently added at a final concentration of 3 mM (ratio IAA: DTG $10:1$) and incubated in the dark for 20 min. All reactions were carried out in methanol at 100 µl final volume. Control samples were included to measure levels of DTG and alkylated gliotoxin formed in the absence of Zn^{2+} . Reactions were evaluated by RP-HPLC.¹⁵

A. fumigatus phenotypic assays

A. fumigatus Δ gliT and Δ gliT:: Δ gtmA strains were grown on MEA agar for 5 days at 37 \degree C. After incubation, conidia were harvested with PBST and washed three times with PBS and resuspended in PBS. Conidia were counted using haemocytometer and stored at 4 °C for future use. Conidia were serially diluted to 10^3 μ l⁻¹ and 5 ml was spotted on Czapek-Dox agar plates (permissive for endogenous gliotoxin biosynthesis⁷) containing gliotoxin (0, 15 or 30 μ M) and Zn²⁺, respectively. Plates were then incubated at 37 \degree C and growth monitored up to 96 h by measuring radial growth (mm) of each colony. Two-way ANOVA analysis was performed to determine the statistical significance between strains at different concentrations of gliotoxin and Zn^{2+} .

Extraction and measurement of extracellular and intracellular gliotoxin

A. fumigatus wild-type and $\Delta gliT$:: Δg tmA were grown in Sabourauddextrose media for 21 h and gliotoxin was subsequently added for 3 h at 15 µM with and without Zn^{2+} (*n* = 3 biological replicates for all specimens). Zn^{2+} was added at 1 mM final concentration. Supernatant samples were taken after 15, 30, 60, 120 and 180 min and were extracted using chloroform $(1:1)$, as described.³¹ The organic extracts were subsequently dried down and resuspended in methanol and analysed for gliotoxin content using RP-HPLC with UV detection (Shimadzu), using polar C18 RP-HPLC column (Phenomenex polar C18 Luna Omega column (150 mm 4.6 mm, 5 μ m)) at a flow rate of 1 ml min⁻¹. A mobile phase of acetonitrile and water with 0.1% (v/v) TFA was used under gradient conditions. A. fumigatus wild-type was also grown in Czapek-Dox media for 72 h (gliotoxin-producing conditions) in the presence of either low (0.027 mM) or high (0.5 mM) Zn^{2+} (n = 3 biological replicates for all specimens). Controls were included, with no Zn^{2+} added. Aliquots were taken every 24 h and organic extractions and RP-HPLC analyses were performed as outlined above. For intracellular gliotoxin recovery, mycelia collected after 3 h incubation were harvested and snap frozen in liquid N2. Intracellular gliotoxin was extracted as described previously for SAM, using a modified protocol.⁸ Briefly, mycelia were ground using liquid N_2 with a mortar and pestle. 100 mg mycelia were incubated with 0.1 N HCl (250 μ l) on ice for 1 h with intermittent vortexing. Protein was removed by addition of 100% TCA to achieve a final concentration of 15% (v/v) TCA. After centrifugation at 16 000 \times g, supernatants were collected and then analysed by RP-HPLC.

Proteomic analysis of gliotoxin affects, with and without Zn^{2+} , on A. fumigatus Δ gliT

A. fumigatus Δ gliT was cultured (n = 3/condition) for 21 h in Sabouraud-dextrose media (SDM) followed by addition of gliotoxin (15 μ M) or methanol (control) with and without ZnSO₄ (1 mM) addition for 3 h. Mycelia were then harvested and snap frozen in liquid N_2 . Mycelia were lysed using buffer (100 mM Tris–HCl, 50 mM NaCl, 20 mM EDTA, 10% v/v glycerol. 1 mM PMSF, 1 μ g ml⁻¹ pepstatin A, pH 7.5) with grinding and sonication, and clarified using centrifugation.⁸ The protein lysates were then subjected to precipitation using TCA/acetone and resuspended in 100 mM Tris–HCl, 6 M urea, 2 M thiourea, pH 8.0. Samples were then reduced by DTT and followed by alkylation using IAA. This was followed by addition of sequencing grade trypsin combined with Protease Max surfactant for overnight digestion. Following sample clean-up using Millipore C18 ZipTips, all peptide mixtures were analysed via a Thermo-Fisher Q-Exactive mass spectrometer coupled to a Dionex RSLCnano.³² LC gradients ran from 4% to 35% ACN over 2 h, and data were collected using a Top15 method for MS/MS scans. Comparative proteome abundance and data analysis was performed using MaxQuant software (v. 1.5.3.30), with PERSEUS used to organize data (v. $1.5.4.0$).^{33,34} Proteins with significant changes in abundance ($p < 0.05$; fold change ≥ 2) signified quantitative changes between the sample treatments. Qualitative differences were also noted, where proteins were not detected in one group and were present in at least 2 replicates from the comparator group. Functional analysis was carried out using FungiFun2 ([https://](https://elbe) elbe.hki-jena.de/fungifun/fungifun.php).³⁵ Paper

Version Microsoften article on 2018 2018. These band on the second of the second of the second on the control of the second on 2018. The second of the second on 2018. The second of the second on the second on the s

In vitro DTG induced-protein aggregation assays

Protein extracts were prepared from A. fumigatus mycelia from three biological replicates, previously cultured in SDM at 37 $^{\circ}$ C, for 24 h at 200 rpm. Mycelia were harvested by filtering through Miracloth, snap-frozen and ground in liquid N_2 using a pestle and mortar. Lysis buffer (0.1 M Tris–HCl, 50 mM NaCl, 1 mM PMSF pH 7.5; 400 µl) was added to ground mycelia (200 mg) and lysed by sonication. Lysates were clarified by centrifugation at 12000g, 4 \degree C for 10 min and this was repeated twice to remove insoluble material. Lysates were diluted to 1 mg ml^{-1} and subjected to the following treatments in triplicate: (a) control: methanol and TCEP were added as a solvent control; (b) DTG: TCEP-reduced gliotoxin was added to the protein extracts to a final concentration of 150 μ M and (c) DTG/Zn²⁺:TCEP-reduced gliotoxin was pre-mixed with $ZnSO₄·7H₂O$ at a 1:3 molar ratio. The mixture was added to the protein extracts in triplicate to achieve a final concentration of 150 μ M DTG and 450 μ M Zn²⁺. All samples were incubated at 50 $^{\circ}$ C for 30 min prior to addition of solubilisation buffer and analysis by SDS-PAGE. Protein aggregates were observed as a band of Coomassie-stained protein at the interface between the stacking and resolving

gels. These bands were excised and subjected to in-gel digestion according to Shevchenko et al.³⁶ Peptide mixtures were de-salted using C_{18} Zip-tips and analysed using Label-free quantitative proteomics (Thermo Q-Exactive LC-MS). Data analysis was performed using MaxQuant with Perseus for data organisation and statistical analysis.^{9,32}

Results

DTG chelates Zn^{2+} in vitro

High resolution mass spectrometry, spectrophotometry, and chemical and enzyme assays were used to investigate the formation of a complex between DTG and Zn^{2+} . Following TCEP-mediated reduction of gliotoxin and Zn^{2+} addition, negative mode MS analysis revealed the presence of a Zn[dithiolate gliotoxin] complex with $m/z = 424.93851$, corresponding to a Cl⁻ adduct of the metal chelate (Fig. 2A and Fig. S1, ESI[†]). Furthermore, DTG also competitively dissociated Zn^{2+} from a $Zn[(4-(2-pyridylazo)-resorcinol)_2]$ $Zn(PAR)_2$ complex, whereby DTG (60-300 μ M) addition resulted in a corresponding decrease in A_{493nm} of $\text{Zn}(\text{PAR})_2$, corresponding to Zn^{2+} chelation (Fig. 2B). Relevantly, gliotoxin had no effect on $Zn(PAR)$ ₂ interaction which confirmed the specific role of the dithiol moiety of reduced gliotoxin in chelation (Fig. S2, ESI†). To investigate the effect of Zn^{2+} on the enzymatic S-methylation of DTG, a GtmA activity assay was performed. GtmA bis-thiomethylates DTG to form BmGT using SAM as the methyl source (Fig. 1).⁹ Gliotoxin S-methylation occurs in a sequential manner, with mono-methylated species (MmGT) appearing initially, followed by formation of BmGT. 9,15,37 Zn²⁺ inhibits GtmA-mediated thiomethylation of gliotoxin in vitro in a concentration-dependent manner, most likely due to an inability to recognise and bind $Zn^{2+}[DTG]$ (Fig. 2C). Support for DTG thiol protection by Zn^{2+} coordination was provided by an alkylation assay. Although IAA modifies DTG to alkylated gliotoxin, Zn^{2+} also inhibits alkylation of DTG which further strengthens the concept of thiol occlusion via Zn^{2+} chelation (Fig. 2D and Fig. S3, ESI[†]).

DTG inhibits AP by Zn^{2+} chelation

To investigate the ability of DTG to inhibit the activity of a zincdependent metalloenzyme, AP activity assays were carried out. Gliotoxin was reduced to DTG as described above and confirmed by RP-HPLC (data not shown). DTG inhibited AP in a dosedependent manner (0-150 μ M DTG) and time-dependent manner (Fig. 3A and Fig. S4, ESI†). Solvent controls (containing TCEP and methanol) did not show any significant AP inhibition. While 50 µM DTG significantly inhibited AP activity ($p < 0.001$), preincubation of DTG (50 μ M) with Zn²⁺ (25 μ M–1 mM) alleviated DTG-associated AP inhibition (Fig. 3B). The Zn^{2+} chelator TPEN was also tested, but only produced a minor decrease in AP activity (11%) when used at the same concentration as DTG (50 μ M) (Fig. 3C). EDTA (5 mM), a metal ion chelator, inhibited AP activity by approx. 30–40% (Fig. S5, ESI†). AP activity was not significantly inhibited by 150 μ M GT, 50 μ M-5 mM of the cellular reductant GSH or 50 μ M-1 mM of the thiol-containing reductant DTT (Fig. S5, ESI†). Lack of inhibition by GT, TCEP, GSH or DTT

Fig. 2 Characterisation of DTG as a Zn²⁺ chelator. (A) Dithiolate gliotoxin complexed with Zn²⁺ (Cl⁻ adduct) with accompanying isotopic analysis in negative ion mode MS (direct injection). Structure of DTG:Zn complex is depicted with positioning of the metal ion between the two thiolates, although exact coordination with other groups is unknown. (B) DTG chelates Zn^{2+} from $Zn(PAR)_{2}$ (Zn(PAR)₂ pK_d = 12.15)⁴⁴ in a concentration-dependent manner. Overlaid UV-vis profiles of PAR alone (I; red), and Zn(PAR)₂ (II; blue). DTG added in increased concentration to Zn(PAR)₂ 1–5 molar equivalents of oxidised gliotoxin to Zn^{2+} ; III–VII, respectively. Decrease in A_{493 nm} is accompanied by increase in A_{413 nm} as free PAR is released. (C) Zn^{2+} inhibits GtmA-mediated methylation of DTG in a concentration dependent manner. Zn^{2+} was included in the methylation assay at a molar ratio ranging from 0.0001 to 10:1 (DTG). A control containing no Zn^{2+} was included. Concentration (μ M) of BmGT and MmGT formed after 5 min was determined by RP-HPLC. (D) Zn^{2+} inhibits IAA-mediated alkylation of DTG in a concentration-dependent manner. RP-HPLC analysis (detection at 254 nm) reveals the inhibitory effect of Zn²⁺ on DTG alkylation.

strongly infers that inhibition is not caused by AP disulphide bridge cleavage or thiol modification. Zinc addition for 30 min

reverses DTG-induced inhibition of AP, which further implicates Zn^{2+} chelation as the mechanism of inhibition (Fig. S6, ESI†).

Fig. 3 DTG inhibits AP activity by Zn^{2+} chelation. (A) Dose-dependent (10–150 μ M DTG) inhibition of AP activity, with IC₅₀ = 38 μ M DTG. Neither GT or the solvent control (12.5 mM TCEP in methanol) inhibit AP activity. (B) Pre-incubation of DTG with Zn^{2+} (0.1, 0.5 and 1 mM) prevents DTGmediated enzyme inhibition. (C) TPEN (50 μ M) has no significant impact on AP activity (89% activity compared to control). Pre-incubation of DTG with Zn^{2+} (25–50 µM) significantly recovered AP activity to 86% of the control.

Impact of Zn^{2+} exposure on A. fumigatus $\Delta gliT$ and $\Delta gliT$:: $\Delta gtmA$, in the presence of exogenous gliotoxin

In order to determine whether Zn^{2+} had an impact on gliotoxinassociated growth inhibition, plate assays were carried out using A. fumigatus strains deficient in enzymes that use DTG as their substrate, $\Delta gliT$:: $\Delta gtmA$ and $\Delta gliT$. Severe radial growth inhibition of A. fumigatus $\Delta g l i T$:: $\Delta g t m A$ in the presence of gliotoxin (15 μ M)¹⁵ (Fig. 4A) was significantly relieved ($p <$ 0.005) in the presence of 1–10 μ M Zn²⁺, and relief of gliotoxinmediated growth inhibition was maintained up to 50 μ M Zn²⁺. Radial growth of A. fumigatus Δ gliT:: Δ gtmA was unaffected by Zn^{2+} (1–50 µM). Higher concentrations of Zn^{2+} significantly inhibited the growth of both A. fumigatus $\Delta g l i T$: $\Delta g t m A$ (100 µM-1 mM; Fig. 4) and A. fumigatus $\Delta g \, l \, l$ (1–2 mM; Fig. S7, ESI†). This correlates with a decrease in the rescue effect of Zn^{2+} on gliotoxinassociated growth inhibition in A. fumigatus $\Delta g l i T$:: $\Delta g t m A$. The combination of gliotoxin (15–30 μ M) and excess Zn²⁺ (1–2 mM) produced further inhibition of radial growth of A. fumigatus $\Delta gliT$ at 96 h ($p < 0.005$) compared to the growth reduction observed in either the presence of Zn^{2+} or gliotoxin alone (Fig. S7, ESI†).

Fig. 4 (A) Potent growth inhibitory effect of gliotoxin (15 μ M) on A. fumigatus Δ gliT:: Δ gtmA is significantly (p < 0.005) reversed by co-addition of Zn²⁺ (1-50 μ M). Zn²⁺ (0-50 μ M) has a negligible effect on mutant growth. (B) Graphical representation of same in terms of relative growth percentages. Note that higher Zn^{2+} levels (100 μ M-1 mM) result in dissipation of the relief, in line with Zn^{2+} -associated toxicity noted in controls.

Zn^{2+} augments gliotoxin uptake and efflux in A. fumigatus wild-type and $\Delta gliT$:: $\Delta gtmA$

Previous studies have revealed that exogenous gliotoxin is sequestered by A. fumigatus.³⁸ To further explore the interaction between Zn^{2+} and gliotoxin *in vivo*, both *A. fumigatus* wild-type and $\Delta gliT$: $\Delta gtmA$ were exposed to gliotoxin (15 µM) for 3 h in the presence and absence of Zn^{2+} (1 mM). In all conditions uptake of gliotoxin by A. fumigatus was evident by decreasing levels detected in culture supernatants across the incubation times, and corresponding increases in intracellular gliotoxin concentration (Fig. S8, ESI†). Fig. S8A–D (ESI†) show that excess Zn^{2+} augmented gliotoxin uptake in both A. *fumigatus* wild-type and $\Delta gliT$:: $\Delta gtmA$. In the presence of Zn^{2+} (1 mM), with significantly less gliotoxin detected in A. fumigatus wild-type culture supernatants 30 min after addition (Fig. S8A–C; $p < 0.05$, ESI[†]). Zn²⁺ effected a more marked response in A. fumigatus Δ gliT:: Δ gtmA, with a significant decrease in extracellular gliotoxin levels at every time point (15–180 min; Fig. S8B; $p < 0.05$, ESI†). When excess Zn²⁺ was present in the culture medium, initial rates of gliotoxin uptake increased from 38.32 \pm 5 to 55.04 \pm 8.78 ng min $^{-1}$ ml $^{-1}$ of culture (mean \pm SEM) in A. fumigatus wild-type and from 5.06 ± 7.16 to 75.94 ± 7.38 ng min $^{-1}$ ml $^{-1}$ of culture (mean \pm SEM)

Metallomics Paper

in $\Delta gliT$: $\Delta gtmA$. Commensurate increases in intracellular gliotoxin (ng/100 mg mycelia) were observed in both A. fumigatus wildtype and $\Delta gliT$:: $\Delta gtmA$, in the presence of Zn^{2+} (Fig. S8D, ESI†). Intracellular levels of gliotoxin were significantly increased in A. *fumigatus* wild-type upon Zn^{2+} co-addition ($p < 0.01$), likely due to inhibition of GtmA-catalysed BmGT formation and efflux. Elevated intracellular gliotoxin levels in $\Delta g l i T$:: $\Delta g t m A$ correlated with an increased rate of gliotoxin uptake induced by Zn^{2+} (Fig. S8, ESI†). Intracellular DTG was not detectable due to either spontaneous intracellular or extraction-associated oxidation.

LFQ proteomic analysis reveals that Zn^{2+} alters the effect of gliotoxin on A. fumigatus Δ gliT

To further investigate the nature of the combinatorial inhibition of A. fumigatus $\Delta g l \bar{l} T$ during simultaneous exposure to $\rm Zn^{2+}$ and gliotoxin, compared to gliotoxin alone, comparative LFQ proteomic analysis was carried out. Overall, 47 proteins exhibited unique presence and 23 proteins were significantly increased in abundance in A. fumigatus Δg liT upon combinatorial exposure (Zn^{2+}/GT) compared to gliotoxin only (Table S1, ESI†). Conversely, 45 proteins were absent and 10 proteins showed significantly reduced abundance in A. fumigatus $\Delta g l i T$ upon combinatorial exposure compared to gliotoxin only (Table S1, ESI†). Functional analysis revealed that zinc-ion binding proteins were significantly enriched (8/70 proteins; $p = 0.005$) among proteins with elevated abundance or uniquely detected in the presence of Zn^{2+}/GT compared to gliotoxin alone (Table 1 and Fig. S9, ESI†). Notably, secondary metabolism-associated proteins were altered in abundance, or demonstrated qualitative changes in the combined condition (Zn^{2+}/GT) compared to gliotoxin only (Table 2). These included the ferricrocin synthetase SidC, which was repressed by gliotoxin, but not by the combined condition, as were four members of the fumagillin biosynthetic cluster (Table 2 and Table S1, ESI†). Consequently, when A. fumigatus was grown in gliotoxin-producing conditions, Zn^{2+} caused a significant increase in fumagillin levels even at lower levels $(0.027 \text{ mM } \text{Zn}^{2+}; p < 0.001)$ (Fig. S10, ESI†). Metaleonics

Severate comparison control and the present of the Access Article. Published on 10:31 Article is a control and the common points of the dependent of the second of the second on 10 are the second on 10:31 Arti

Low affinity zinc transporter ZrfB, which is ZafA-dependent and normally induced by zinc depletion, was undetectable in the combinatorial exposure condition, which suggests a zinclimiting environment in gliotoxin-only exposure (Table 1). Likewise, allergen AspF2 which is expressed under zinc-limiting conditions, and shares a divergent promoter with z rf C , was absent upon co-exposure. 39 It was also observed that GliM, a component of the gliotoxin biosynthetic capacity was absent, which infers that gliotoxin biosynthesis may be attenuated in the combinatorial exposure scenario (Table 2). Indeed, Zn^{2+} presence completely abrogated gliotoxin biosynthesis in Czapek-Dox media (Fig. S10, ESI†). These observations pertaining to impeded gliotoxin biosynthesis are further supported by the absence of GpgA (AFUA_1G05210), a G-protein coupled receptor subunit, previously shown to be essential for gliotoxin production in A. fumigatus.⁴⁰ It is also notable that the nonribosomal peptide synthetase NRPS8/Pes 3^{41} (AFUA_5G12730) was absent following co-exposure to Zn^{2+} and gliotoxin, which suggests that it may contribute to adaptation to zinc-limiting conditions.

Gliotoxin exposure induces elevated GtmA abundance in A. $\emph{fumigatus}$ $\Delta \emph{gliT},^9$ and Manzanares-Miralles *et al*. reported that gliotoxin induces significantly increased abundance of the GtmA ortholog in Aspergillus niger, in which the gli cluster is absent.³⁷ Interestingly, while Zn^{2+} exposure alone had no significant effect on the abundance of GtmA in A. fumigatus, combinatorial exposure with gliotoxin resulted in significantly elevated GtmA abundance (18.8-fold and $p < 0.01$) compared to that in the presence of gliotoxin only (Table 2). Allied to GtmA activity data (Fig. 2C), this suggests that in vivo chelation of Zn^{2+} by DTG may prevent substrate access to GtmA, thereby resulting in a further increased abundance of the enzyme (Fig. 5A).

Manifestation of heat-induced in vitro protein aggregation in A. fumigatus mycelial lysates, specifically by DTG

Inhibition of Zn^{2+} -dependent enzyme activities, combined with the observed in vivo interplay between the divalent metal cation

Table 1 Zinc associated proteins with differential abundance, or uniquely detected, between the combination condition (Zn^{2+}/GT) and the gliotoxin – only control

^a Negative values indicate protein was decreased in abundance in combination treatment (Zn^{2+}/GT) compared to the gliotoxin only treatment, while positive values indicate an increase in abundance. ^b Uniquely detected: protein uniquely identified in at least 2 biological replicates of either gliotoxin-only treated cultures or the combination condition $(2n^{2+}/GT)$.

Table 2 Secondary metabolism-related proteins with differential abundance, or uniquely detected, between the combination condition (Zn^{2+}/GT) and the gliotoxin-only control

Secondary metabolite	Protein description	log ₂ fold increase $(p$ value)	Uniquely detected ^a	Peptides	Gene IDs $(AFUA_$
Gliotoxin	Unknown function protein GliH		GT	3	6G09745
	O-Methyltransferase GliM		GT	4	6G09680
	Gliotoxin thiomethyltransferase GtmA	4.2361(0.0052)		18	2G11120
NRPS8 (unknown)	NRPS8/Pes3		GT	3	5G12730
Ferricrocin	NRPS involved in ferricrocin siderophore biosynthesis SidC		Zn^{2+}/GT	5	1G17200
Fumagillin	Putative alpha/beta hydrolase FmaC		Zn^{2+}/GT Zn^{2+}/GT	11	8G00380
	Unknown function protein FmaD Protein encoded in the fumagillin gene cluster		Zn^{2+}/GT	5 10	8G00400 8G00430
	Putative iron-dependent oxygenase FmaF		Zn^{2+}/GT	9	8G00480
	and DTG in A. fumigatus, led us to further investigate the inter- action between DTG and A. fumigatus proteins. DTG-associated protein destabilisation was investigated by exposing complex	spectrometry-based analysis of the proteins retained at the interface between the stacking and resolving gel identified			
	lysates to either DTG or a mixture of Zn ²⁺ and DTG, followed by heat treatment. Exposure of A. fumigatus mycelial protein lysates to	551 proteins in total from the three conditions tested. Label-free proteomics revealed 94 proteins uniquely detected or significantly			
	TCEP/methanol, DTG or a mixture of Zn^{2+} and DTG followed by	elevated in the aggregates following DTG pre-treatment, relative to			disaggregate unfolded proteins was observed (Fig. 5B). Mass
	SDS-PAGE analysis revealed near-identical electrophoretic patterns	the control (Fig. 5C and Table S2, ESI ⁺). DTG addition to			
	in the resolving gel (Fig. 5B). However, uniquely, following protein lysate exposure to DTG, and subsequent heat treatment	A. fumigatus lysates elicits an overall increase in aggregated pro- teins compared to the controls. Functional analysis revealed a			
	(50 °C/30 min), either protein aggregation or an inability to	significant number of proteins with oxidoreductase activity			
		$(n = 27, p = 0.019)$, metal ion binding $(n = 16, p = 0.024)$ and			
		zinc ion binding ($n = 7$, $p = 0.047$) showed increased aggregation			
Α.	В. DTG-treated DTG:Zn ²⁺ (1:3)- Control treated	in response to DTG treatment (Fig. 5D). Notably, pre-incubation			
*** 500 $**$		of DTG with Zn^{2+} prevented this aggregation in the case of 77%			
	kDa	of the DTG-responsive proteins (72/94 proteins), whereby the			
400	130	combinatorial condition lead to significantly lower levels of			
	$70 -$	protein aggregation compared to DTG treatment ($p < 0.05$).			
		Additional stabilisation was observed in the combinatorial condition,			
GtmA Fold Change ခွဲ မွိ 100		as 77 proteins showed lower abundance in the aggregates compared			

Fig. 5 (A) Fold change in GtmA abundance following gliotoxin (GT: 15μ M) and gliotoxin & Zn^{2+} (15 µM & 1 mM, respectively) exposure compared to controls in A. fumigatus Δ gliT. (B) SDS-PAGE analysis of 3 independently prepared A. fumigatus protein lysates. Lanes 1–3: negative controls: protein lysates treated with TCEP/methanol only. Lanes 4–6: protein lysates treated with DTG. Lanes 7–9: protein lysates treated with DTG : Zn^{2+} (1 : 3). M, molecular mass marker. Protein aggregation is evident (boxes) following DTG-exposure. (C) Number of proteins with increased or decreased abundance (qualitative and quantitative) in aggregates recovered from control, DTG or combined DTG/Zn^{2+} treatments. (D) Functional analysis (FungiFun2) showing GO molecular function categories significantly enriched ($p < 0.05$) among proteins with increased abundance in DTGinduced aggregates. Arranged in order of ascending p value.

disaggregate unfolded proteins was observed (Fig. 5B). Mass spectrometry-based analysis of the proteins retained at the interface between the stacking and resolving gel identified 551 proteins in total from the three conditions tested. Label-free proteomics revealed 94 proteins uniquely detected or significantly elevated in the aggregates following DTG pre-treatment, relative to the control (Fig. 5C and Table S2, ESI†). DTG addition to A. fumigatus lysates elicits an overall increase in aggregated proteins compared to the controls. Functional analysis revealed a significant number of proteins with oxidoreductase activity $(n = 27, p = 0.019)$, metal ion binding $(n = 16, p = 0.024)$ and zinc ion binding ($n = 7$, $p = 0.047$) showed increased aggregation in response to DTG treatment (Fig. 5D). Notably, pre-incubation of DTG with Zn^{2+} prevented this aggregation in the case of 77% of the DTG-responsive proteins (72/94 proteins), whereby the combinatorial condition lead to significantly lower levels of protein aggregation compared to DTG treatment $(p < 0.05)$. Additional stabilisation was observed in the combinatorial condition, as 77 proteins showed lower abundance in the aggregates compared to the control condition (Fig. 5C). Excess Zn^{2+} used in the pre-mixed DTG/Zn^{2+} treatment could contribute to this additional stabilising effect.

Among the proteins affected by DTG-induced aggregation, was the zinc-dependent alcohol dehydrogenase AlcC, which was only detected in aggregates from lysates pre-incubated with DTG, indicating co-addition of Zn^{2+} prevented DTG-induced aggregation of this protein. Another protein detected exclusively in aggregates induced by DTG was the putative farnesyltransferase beta subunit Ram1 (AFUA_4G10330) which contains a C-terminal Zn^{2+} binding pocket, within the active site.^{42,43} A number metallopeptidases were also increased or uniquely detected in DTGinduced aggregates (AFUA_6G09190, AFUA_4G07910, MepB), as well as the zinc-dependent methionine synthase MetH/D, while co-addition of Zn^{2+} generated aggregation profiles in line with the control.

Overall, these data suggest that DTG may chelate Zn^{2+} from selected cellular proteins thereby inducing protein unfolding and consequently resulting in temperature-induced aggregation. It is also possible that DTG inhibits selected components of the proteasome-mediated recognition and/or digestion of protein aggregates; which results in their persistence in DTG-treated lysates. Finally, it is likely that some proteins detected in the aggregates may be present as a result of non-specific physical entrapment, however, these putative, non-specifically aggregated proteins are resistant to the dissolution effects of SDS solubilisation buffer and heat (95 $°C/4$ min).

Discussion

Herein, we provide evidence that DTG is a Zn^{2+} -chelator. We also reveal Zn^{2+} -metalloenzyme inhibition specifically by DTG, and not by gliotoxin or other reducing agents (DTT or GSH). This suggests Zn^{2+} chelation, as well as disulphide bridge cleavage or thiol modification, as a key mode of enzyme inhibition by DTG. We propose that DTG, acting as a Zn^{2+} -chelator, can significantly inhibit growth of A. fumigatus completely deficient in essential oxidation or thiomethylation activities. Unbiased LFQ proteomics reveals that Zn^{2+} can significantly modify the nature and extent of protein abundance alterations caused by A . fumigatus Δg liT exposure to gliotoxin. This important revelation confirms in vivo interaction between both molecular species. Moreover, additional LFQ proteomic data reveal that in vitro, DTG can cause specific protein aggregation, manifested by heat instability, possibly due to structural alteration to known Zn^{2+} -dependent enzymes. These unexpected observations open a new front in our exploration of the metallo-metabolome in fungi, and possibly other species.

High resolution, negative mode MS revealed that DTG complexes Zn^{2+} and exhibits a monoisotopic peak with m/z 424.93749, which equates to a single $Zn^{2+}[DTG]$ complex detected as a Cl^{-} adduct $[$ (DTG + ⁶⁴Zn)-2H + Cl]⁻. Previous work by Woodcock et al. proposed the existence of a similar adduct $(m/z 427)$, following NaBH₄-mediated reduction.²⁴ Since no spectra were provided, it is not possible to ascertain the origin of the different observed m/z values, although it seems likely that Woodcock et al. reported the average m/z for the compound or the m/z of the base peak $[({\rm DTG + ^{66}\!Zn})$ –2H + Cl] $^-$, rather than the monoisotopic peak. 24 In addition, we have found that DTG efficiently chelates Zn^{2+} from $Zn(PAR)_2$ (60 µM), in a dose-dependent manner from 60 to 300 µM DTG. Given that Kocyła et al. have proposed an effective pK_d of 12.15 at pH 7.4 for $Zn(PAR)₂$, it is clear that the affinity of DTG for Zn^{2+} must equate to, or exceed, this value as Zn^{2+} is removed from the $Zn(PAR)_2$ complex when equimolar amounts of DTG are added to it.⁴⁴ However, it is not ideal to compare these complexes directly due to differences in stoichiometry. Chan et al. also deployed a PARbased assay system, which involved chelation of Zn^{2+} from $Zn(PAR)_{2}$, and subsequent formation of stable complexes to reveal that Zn^2 is chelated by dithiol holomycin (Fig. 1; termed red-holomycin in Chan et al.) with high affinity.³⁰ Davis et al. previously revealed and characterised alkylation of DTG using 5'-iodoacetamidofluorescein.45 In the present study, we observed that IAA-mediated alkylation of DTG was inhibited by Zn^{2+} , which further underpins our proposal that a Zn[DTG] complex is formed via thiolate coordination and is stable under in vitro conditions used.

DTG acts as a potent inhibitor of the Zn^{2+} -dependent enzyme, AP. Previous studies have reported the inhibitory effects of gliotoxin on mammalian Zn^{2+} -dependent enzymes, without attributing this

activity to zinc chelation.^{16,17} Vigushin et al. studied the inhibition of Zn^{2+} -dependent farnesyltransferase (FTase) and geranyltransferase (GGTase) I by gliotoxin, proposing thiol modification of these enzymes as a possible mechanism of action. Importantly, the authors noted that these assays required the reductant DTT, which aligns with our observations of the inhibition of Zn^{2+} -dependent metalloenzymes (i.e., AP) by DTG rather than gliotoxin. Our data also provides an alternative explanation for previous observations which suggested that reducing agents enhanced the inhibitory activity of gliotoxin against a Zn^{2+} -dependent equine ADH.¹⁶ Although these authors posited a redox explanation, it is equally plausible that Zn^{2+} chelation by DTG effected equine ADH inhibition. This accords with our observations that prior incubation of DTG with Zn^{2+} prevents inactivation of AP, and also that DTGassociated AP inhibition can be rescued by subsequent addition of Zn^{2+} (Fig. 3 and Fig. S6, ESI†).

Relevantly, it has been shown that ETPs, in particular gliotoxin, block the interaction between Hypoxia Inducible Factor- α (HIF- α) and the transcriptional coactivator p300 by a Zn^{2+} ejection mechanism.18 Moreover, these authors noted the antiproliferative effects of ETPs, and provide significant insight into their mechanism of action in animal cells. Cook et al. reference the Zn^{2+} -dependency of many gliotoxin-sensitive enzymes and the Zn^{2+} requirement of GliZ, the transcription factor essential for gliotoxin biosynthesis.^{1,18} However, their proposed mechanism of action of Zn^{2+} ejection from p300 by gliotoxin does not take into account the presence of DTT in assay buffers.18 Thus, consequent to our observations with AP and those of Cook et al., we now speculate that it is DTG, formed due to the presence of equivalent amounts of DTT, and not gliotoxin per se, which causes Zn^{2+} ejection from p300. Relevantly, it has recently been elegantly demonstrated that dithiol holomycin (Fig. 1) can effect Zn^{2+} chelation and cause inhibition of a metallo- β -lactamase.³⁰ Our demonstration of DTG-mediated inhibition of AP suggests that the Zn^{2+} chelation potential of dithiolopyrrolones, like holomycin, extends to ETPs. Metaleonics

Useasce, Finally, it is likely that some proteins detected in the artivity one chelsion,¹⁶³ Vapakin *red*, and generation are a consider an experiment and the commence of the solution-of the solution-of the

> Our deployment of A. fumigatus mutants deficient in gliotoxin self-protection $(\Delta gliT)$ and self-protection/negative regulation of gliotoxin biosynthesis in combination $(\Delta g l i T:: \Delta g t m A)$ reveal a hitherto unknown systems interaction between Zn^{2+} and gliotoxin biochemistry. We have observed the in vitro inhibition of GtmAmediated BmGT formation by Zn^{2+} . This mode of inhibition appears to result from DTG chelation of Zn^{2+} , and its subsequent unavailability as a substrate to GtmA, as opposed to direct enzyme inhibition. Interestingly, this observation is in accordance with the effect of Zn^{2+} (1 and 2 mM) exposure towards A. fumigatus $\Delta gliT$, whereby combinatorial gliotoxin/ Zn^{2+} exposure significantly augments growth inhibition. In effect, we postulate that elevated levels of Zn^{2+} may augment intracellular Zn(DTG) complex formation, resulting in non-availability to GtmA, which actually significantly increases GtmA abundance (Fig. 5A). Thus, we speculate that Zn^{2+} -mediated disruption of GtmA functionality in A. fumigatus Δ gliT partially creates an inability to dissipate DTG, though not as absolute as pertains in A. fumigatus Δg liT:: Δg tmA. Indeed, A. fumigatus Δ gliT:: Δ gtmA presents an ideal system to explore the functionality of Zn^{2+} -dependent systems in fungi, since the

Fig. 6 Overview of the proposed mechanism and consequences of Zn^{2+} chelation by intracellular DTG in A. fumigatus. DTG is produced consequent to gliotoxin uptake, via GSH-mediated chemical reduction, or de novo gliotoxin biosynthesis. Under normal conditions, two enzyme systems, namely GliT-mediated oxidation and GtmA-mediated thiomethylation effect dismutation of DTG to prevent interference with intracellular zinc homeostasis. In the absence of both enzymes, elevated intracellular DTG can either chelate free Zn^{2+} or chelate it from intracellular Zn^{2+} -dependent metalloenzymes, causing extreme growth inhibition. Additionally, Zn(DTG) cannot be converted to BmGT via GtmA. Either way, disruption of intracellular zinc homeostasis occurs which leads to activation of zinc uptake systems, potentially via ZafA induction of ZrfA-C expression.

organism lacks both enzymes, GliT and GtmA, which contribute to dissipation of intracellular DTG (Fig. 6), ultimately via either gliotoxin or BmGT efflux.15 The significant, though incomplete, Zn^{2+} -mediated reversal of A. fumigatus Δ gliT:: Δ gtmA sensitivity to exogenous gliotoxin exposure implicates either chelation of free Zn^{2+} or chelation of Zn^{2+} from cellular metalloenzymes, by DTG as key inhibitory mechanisms (Fig. 6). To our knowledge, this is the first report of an in vivo interaction between gliotoxin biosynthesis, dysregulated DTG presence and growth inhibition due to potential interference with Zn^{2+} -associated growth systems. Moreover, it contributes to explaining the cryptic observation of Dolan et al. that A. fumigatus $\Delta g l i T$: $\Delta g t m A$ is the most gliotoxinsensitive mutant observed to date, 15 and also why there are two distinct enzymatic activities which can prevent intracellular DTG accumulation. Overall, these data indicate the inhibitory potential of this endogenous, and potent, Zn^{2+} chelator in A. fumigatus in particular, and possibly microorganisms in general. Indeed, gliotoxin significantly inhibits growth of a range of fungi, $12,46$ and previously work has also indicated its potent anti-bacterial activity.⁴⁷

It is clear that interference with intracellular DTG levels, via gliotoxin exposure to A. fumigatus $\Delta g \, l \, l$ impacts on the fungal proteome, which in turn can be modulated by Zn^{2+} presence. Indeed, it has been estimated that up to 6% (600/10 000) of the A. fumigatus proteome comprises Zn^{2+} -binding proteins; interestingly, this includes a prediction of 300 Zn^{2+} finger transcription factors.⁴⁸ Relevantly, co-addition of gliotoxin and Zn^{2+} induced key alterations to abundance of proteins involved in secondary metabolism. Proteins from the gliotoxin biosynthetic cluster, GliM and GliH, were uniquely detected in

mycelia exposed to gliotoxin, indicating that combinatorial exposure with Zn^{2+} prevents activation of gliotoxin cluster expression. While GliM and GliH are induced in response to gliotoxin, these proteins are not detected following BmGT exposure.⁹ This may implicate Zn^{2+} chelation in the activation of the gliotoxin transcription factor GliZ and the induction of cluster expression. O'Keeffe et al. noted gliotoxin exposure suppresses fumagillin cluster expression in A. fumigatus $\Delta g l i T$.¹⁴ This result was reflected in the current study, however co-exposure to gliotoxin and Zn^{2+} reversed this effect, leading to unique detection of fumagillin biosynthetic proteins. As with gliotoxin, the fumagillin cluster is also regulated by a $(2n^{2+})$ ₂Cys₆ transcription factor (FapR/FumR),^{49,50} possibly implicating Zn^{2+} chelation in cluster repression, since exogenous Zn^{2+} blocks gliotoxinassociated repression. These proteomic observations are supported by the switch from gliotoxin to fumagillin production in A. fumigatus upon Zn^{2+} supplementation (Fig. S10, ESI†).

Incubation of A. fumigatus protein lysates with DTG prior to heat treatment resulted in increased protein aggregation, compared to the solvent control. Oxidoreductases, metal-binding proteins and zinc-binding proteins were significantly enriched amongst these DTG-affected proteins ($p < 0.05$). DTG-induced aggregation of zinc-binding proteins including two alcohol dehydrogenases, AlcC and AFUA_1G04620, and the farnesyltransferase (AFUA_ 4G10330) was observed. This is in line with observed inhibition of mammalian homologs of these proteins by gliotoxin in reducing conditions.^{16,17} AlcC was also observed to increase in abundance in vivo when A. fumigatus Δ gliT was treated with a combination of gliotoxin and Zn^{2+} compared to gliotoxin alone (Table S1, ESI†), possibly to compensate for loss of activity. AlcC has been identified as the primary hypoxia-responsive ADH in A. fumigatus, with a potential role in pathogenesis, 51 while farnesyltransferase has a role in signalling and also contributes to disease.⁵² Further investigation could elucidate the effect of DTG on the functionality of these proteins in A. fumigatus. Pre-incubation of DTG with excess Zn^{2+} abrogated this effect, most likely through formation of Zn(DTG) complexes prior to addition to protein lysates, leading to reduced levels of protein aggregation. A number of metallopeptidases (AFUA_6G09190, AFUA_4G07910, AFUA_1G14920 and MepB) were also observed to undergo increased aggregation in response to DTG, while this was prevented in the presence of Zn^{2+} . These peptidases contribute to protein modification and degradation, and so alteration of these processes by DTG has the potential to disrupt protein turnover. Interestingly, gliotoxin has previously been implicated in inhibition of proteolytic activity of human and toxoplasma proteasomes.53,54 The cobalamin-independent methionine synthase MetH/D showed a similar response to DTG treatment, with Zn^{2+} co-addition blocking aggregation. MetH/D contains a zinc-binding site, required for binding and activation of its substrate homocysteine.⁵⁵ DTG-mediated aggregation of MetH/D would interrupt an integral part of primary metabolism and potentially affect pathogenesis.⁵⁶ Interestingly, while metH/D showed no response to gliotoxin in A. fumigatus wild-type, expression was significantly induced in A. fumigatus $\Delta g l i T$ in response to gliotoxin.¹⁴ Persistence of intracellular DTG, in the Paper Access Article on 1000 Microsoft Common Common Care is licensed at the common Care is licensed and the common Care is licensed and the common Care is an access Article is an access Article is an access Article is an

Metallomics Paper

absence of GliT-mediated oxidation, results in disruption of the methionine cycle, with SAM depletion caused by dysregulation of gliotoxin methylation.8 Added to the extensive SAM consumption, DTG-associated destabilisation of methionine synthase through Zn^{2+} -chelation could place an additional strain on the methionine cycle in A. fumigatus Δg liT. While further studies are required to confirm if these proteins are directly affected by DTG in vivo, these results present strong targets for future investigations to elucidate the systemic effect of DTG on A. fumigatus. Of course, Zn^{2+} chelation is likely not the only mechanism by which DTG exerts its effects on the cell, with previous studies illustrating its potential for thiol modification and redox reactions.57,58 This methodology also provides an unbiased discovery-based mechanism allowing for the identification of putative targets of DTG in other complex systems. Metaleonics

Subsection control and Access Article control and Subsection control and the same of the metaleonic point

priori metaleonic control and Subsection control and the same of the metaleonic control and the same

Interestingly, Müller et al. have shown that allicin (diallyl thiosulphinate) can cause thiol stress and severe growth inhibition in bacteria.⁵⁹ Specifically, Muller *et al.* revealed that allicin induced protein aggregation, likely due to S-allylmercapto protein modification, in crude Escherichia coli cell lysates in a concentrationdependent manner. This is in accordance with our observations of DTG-induced protein aggregation in A. fumigatus protein lysates, although fungal protein destabilisation possibly involves Zn^{2+} chelation mechanism, as opposed to protein modification. Future work will clarify the relative contribution of either mechanism.

As can be seen in Fig. S8 (ESI†), gliotoxin addition to A. fumigatus results in uptake, followed by conversion to intracellular DTG and induction/augmentation of gliotoxin biosynthesis, as previously reported in Owens et al. and Dolan et al.^{8,15} Based on our new observations, we now extend this model and provide a mechanistic link between the intracellular presence of DTG and (i) Zn^{2+} depletion leading to increased ZrfB abundance, as well as (ii) potential Zn^{2+} chelation and destabilisation of metalloenzymes. Interestingly, the membrane-permeable zinc chelator, TPEN, which has been used in $vivo^{60}$ and in A. fumigatus studies,⁶¹ demonstrated lower AP inhibition than DTG (Fig. 3). The dissociation constant (K_d) of Zn(TPEN) has been reported as 6.4×10^{-16} M (pK_d 15.2) at pH 7.4, with the same 1:1 stoichiometry as the $Zn(DTG)$ complex.^{62,63} Future studies will quantify the affinity of DTG for Zn^{2+} however, considering DTG caused significantly greater AP inhibition than TPEN, it is reasonable to conclude that DTG is a better Zn^{2+} chelator than TPEN under the conditions tested. TPEN has been shown to act co-operatively with the antifungal drug caspofungin to significantly improve survival in a mouse model system of Invasive Pulmonary Aspergillosis compared to either caspofungin or TPEN administration alone.⁶⁴ This exciting development is important because exposure to combinations of antifungal drugs, acting in synergy, may address both the development of pathogen resistance and toxicity of high therapeutic levels of either drug alone to the recipient. Although TPEN administration may not have any immediate harmful effects in animals, its safety profile following co-administration with antifungal drugs is unknown. Our observation of DTG as an intracellular Zn^{2+} chelator, ideally positions it as a potential endogenous anti-fungal, especially if strategies to interfere with its enzymatic elimination are elucidated in future research.

Conclusions

Overall, new in vivo and in vitro interactions between Zn^{2+} and DTG, with multiple biological consequences, are revealed. A. fumigatus is a pathogen for which limited therapeutic options exist. Although much recent work has greatly increased the understanding of this and other fungal species, investigation of cryptic interaction between Zn^{2+} and BGC-encoded metabolite functionality needs urgent study. Any new systems identified could not only represent antifungal drug targets, but also inform on the exploitation of BGC-encoded, dithiol-containing, metabolites to restrict Zn^{2+} availability in many microbial species.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

AAS was funded by a Government of Ireland Postdoctoral Fellowship from the Irish Research Council (GOIPD/2015/516). SFD was funded by a SPUR studentship from Maynooth University. Mass spectrometry facilities were funded by Science Foundation Ireland (12/RI/2346 (3)) and the Irish Higher Education Authority.

References

- 1 J. W. Bok, D. Chung, S. A. Balajee, K. A. Marr, D. Andes, K. F. Nielsen, J. C. Frisvad, K. A. Kirby and N. P. Keller, Infect. Immun., 2006, 74, 6761–6768.
- 2 R. A. Owens, G. O'Keeffe, K. A. O'Hanlon, L. Gallagher and S. Doyle, in Human Pathogenic Fungi: Molecular Biology and Pathogenic Mechanisms, ed. D. J. Sullivan and G. P. Moran, Caister Academic Press, 2014, pp. 163–194.
- 3 B. Li, W. J. Wever, C. T. Walsh and A. A. Bowers, Nat. Prod. Rep., 2014, 31, 905–923.
- 4 S. K. Dolan, G. O'Keeffe, G. W. Jones and S. Doyle, Trends Microbiol., 2015, 23, 419–428.
- 5 D. H. Scharf, N. Remme, T. Heinekamp, P. Hortschansky, A. A. Brakhage and C. Hertweck, J. Am. Chem. Soc., 2010, 132, 10136–10141.
- 6 B. Li and C. T. Walsh, Biochemistry, 2011, 50, 4615–4622.
- 7 M. Schrettl, S. Carberry, K. Kavanagh, H. Haas, G. W. Jones, J. O'Brien, A. Nolan, J. Stephens, O. Fenelon and S. Doyle, PLoS Pathog., 2010, 6, e1000952.
- 8 R. A. Owens, G. O'Keeffe, E. B. Smith, S. K. Dolan, S. Hammel, K. J. Sheridan, D. A. Fitzpatrick, T. M. Keane, G. W. Jones and S. Doyle, Eukaryotic Cell, 2015, 14, EC.00055.
- 9 S. K. Dolan, R. A. Owens, G. O'Keeffe, S. Hammel, D. A. Fitzpatrick, G. W. Jones and S. Doyle, Chem. Biol., 2014, 21, 999–1012.
- 10 B. Li, R. R. Forseth, A. A. Bowers, F. C. Schroeder and C. T. Walsh, ChemBioChem, 2012, 1–7.
- 11 P. H. Bernardo, N. Brasch, C. L. L. Chai and P. Waring, J. Biol. Chem., 2003, 278, 46549–46555.
- 12 S. Carberry, E. Molloy, S. Hammel, G. O'Keeffe, G. W. Jones, K. Kavanagh and S. Doyle, Fungal Genet. Biol., 2012, 49, 302–312.
- 13 R. A. Cramer, M. P. Gamcsik, R. M. Brooking, L. K. Najvar, W. R. Kirkpatrick, T. F. Patterson, C. J. Balibar, J. R. Graybill, J. R. Perfect, S. N. Abraham and W. J. Steinbach, Eukaryotic Cell, 2006, 5, 972–980.
- 14 G. O'Keeffe, S. Hammel, R. A. Owens, T. M. Keane, D. A. Fitzpatrick, G. W. Jones and S. Doyle, BMC Genomics, 2014, 15, 894.
- 15 S. K. Dolan, T. Bock, V. Hering, R. A. Owens, G. W. Jones, W. Blankenfeldt and S. Doyle, Open Biol., 2017, 7, 160292.
- 16 P. Waring, A. Sjaarda and Q. H. Lin, Biochem. Pharmacol., 1995, 49, 1195–1201.
- 17 D. M. Vigushin, N. Mirsaidi, G. Brooke, C. Sun, P. Pace, L. Inman, C. J. Moody and R. C. Coombes, Med. Oncol., 2004, 21, 21–30.
- 18 K. M. Cook, S. T. Hilton, J. Mecinovic, W. B. Motherwell, W. D. Figg and C. J. Schofield, J. Biol. Chem., 2009, 284, 26831–26838.
- 19 K. M. Reece, E. D. Richardson, K. M. Cook, T. J. Campbell, S. T. Pisle, A. J. Holly, D. J. Venzon, D. J. Liewehr, C. H. Chau, D. K. Price and W. D. Figg, Mol. Cancer, 2014, 13, 91.
- 20 L. Lauinger, J. Li, A. Shostak, I. A. Cemel, N. Ha, Y. Zhang, P. E. Merkl, S. Obermeyer, N. Stankovic-Valentin, T. Schafmeier, W. J. Wever, A. A. Bowers, K. P. Carter, A. E. Palmer, H. Tschochner, F. Melchior, R. J. Deshaies, M. Brunner and A. Diernfellner, Nat. Chem. Biol., 2017, 13, 709–714. Paper Workerto, E. Molley, S. Hammel, G. Orkerto, G. W. Jones, 22 N. M. Molley, R. A. Orento, D. Michair, S. D. Michair, C. C. Hammel, D. Michair, C. Michair, C.
	- 21 A. Abad, J. V. Fernández-Molina, J. Bikandi, A. Ramírez, J. Margareto, J. Sendino, F. Luis Hernando, J. Pontón, J. Garaizar and A. Rementeria, Rev. Iberoam. Micol., 2010, 27, 155–182.
	- 22 C. Coméra, K. André, J. Laffitte, X. Collet, P. Galtier and I. Maridonneau-Parini, Microbes Infect., 2007, 9, 47–54.
	- 23 J. J. Bennison, R. M. Nottingham, E. L. Key and J. J. Parkins, N. Z. Vet. J., 2010, 58, 201–206.
	- 24 J. C. Woodcock, W. Henderson and C. O. Miles, J. Inorg. Biochem., 2001, 85, 187–199.
	- 25 M. A. Moreno, O. Ibrahim-Granet, R. Vicentefranqueira, J. Amich, P. Ave, F. Leal, J.-P. Latge and J. A. Calera, Mol. Microbiol., 2007, 64, 1182–1197.
	- 26 J. Amich, R. Vicentefranqueira, E. Mellado, A. Ruiz-Carmuega, F. Leal and J. A. Calera, Cell. Microbiol., 2014, 16, 548–564.
	- 27 R. Vicentefranqueira, J. Amich, P. Laskaris, O. Ibrahim-Granet, J. P. Latge, H. Toledo, F. Leal and J. A. Calera, Front. Microbiol., 2015, 6, 160.
	- 28 S. Yasmin, B. Abt, M. Schrettl, T. A. A. Moussa, E. R. Werner and H. Haas, Fungal Genet. Biol., 2009, 46, 707–713.
	- 29 S. Doyle, G. W. Jones and S. K. Dolan, Fungal Biol., 2018, 122, 214–221.
	- 30 A. N. Chan, A. L. Shiver, W. J. Wever, S. Z. A. Razvi, M. F. Traxler and B. Li, Proc. Natl. Acad. Sci. U. S. A., 2017, 114, 2717–2722.
	- 31 E. B. Smith, S. K. Dolan, D. A. Fitzpatrick, S. Doyle and G. W. Jones, Microb. Cell, 2016, 3, 120–125.
- 32 N. M. Moloney, R. A. Owens, P. Meleady, M. Henry, S. K. Dolan, E. Mulvihill, M. Clynes and S. Doyle, J. Proteomics, 2016, 136, 99–111.
- 33 J. Cox, M. Y. Hein, C. A. Luber, I. Paron, N. Nagaraj and M. Mann, Mol. Cell. Proteomics, 2014, 13, 2513–2526.
- 34 S. Tyanova, T. Temu, P. Sinitcyn, A. Carlson, M. Y. Hein, T. Geiger, M. Mann and J. Cox, Nat. Methods, 2016, 13, 731–740.
- 35 S. Priebe, C. Kreisel, F. Horn, R. Guthke and J. Linde, Bioinformatics, 2015, 31, 445–446.
- 36 A. Shevchenko, H. Tomas, J. Havlis, J. V. Olsen and M. Mann, Nat. Protoc., 2007, 1, 2856–2860.
- 37 L. Manzanares-Miralles, Ö. Sarikaya-Bayram, E. B. Smith, S. K. Dolan, Ö. Bayram, G. W. Jones and S. Doyle, *J. Proteomics*, 2016, 131, 149–162.
- 38 L. Gallagher, R. A. Owens, G. O'Keeffe, S. K. Dolan, M. Schrettl, K. Kavanagh, G. Jones and S. Doyle, Eukaryotic Cell, 2012, 11, 1226–1238.
- 39 J. Amich, R. Vicentefranqueira, F. Leal and J. A. Calera, Eukaryotic Cell, 2010, 9, 424–437.
- 40 K.-S. Shin, N.-J. Kwon and J.-H. Yu, Curr. Genet., 2009, 55, 631–641.
- 41 K. A. O'Hanlon, T. Cairns, D. Stack, M. Schrettl, E. M. Bignell, K. Kavanagh, S. M. Miggin, G. O'Keeffe, T. O. Larsen and S. Doyle, Infect. Immun., 2011, 79, 3978–3992.
- 42 M. F. Mabanglo, M. A. Hast, N. B. Lubock, H. W. Hellinga and L. S. Beese, Protein Sci., 2014, 23, 289–301.
- 43 H.-W. Park, S. R. Boduluri, J. F. Moomaw, P. J. Casey and L. S. Beese, Science, 1997, 275, 1800–1805.
- 44 A. Kocyła, A. Pomorski and A. Krezel, J. Inorg. Biochem., 2015, 152, 82–92.
- 45 C. Davis, N. Gordon, S. Murphy, I. Singh, K. Kavanagh, S. Carberry and S. Doyle, Anal. Bioanal. Chem., 2011, 401, 2519–2529.
- 46 J. J. Coleman, S. Ghosh, I. Okoli and E. Mylonakis, PLoS One, 2011, 6, e25321.
- 47 W.-L. Liang, X. Le, H.-J. Li, X.-L. Yang, J.-X. Chen, J. Xu, H.-L. Liu, L.-Y. Wang, K.-T. Wang, K.-C. Hu, D.-P. Yang and W.-J. Lan, Mar. Drugs, 2014, 12, 5657–5676.
- 48 C. C. Staats, L. Kmetzsch, A. Schrank and M. H. Vainstein, Front. Cell. Infect. Microbiol., 2013, 3, 65.
- 49 P. Wiemann, C. Guo, J. M. Palmer, R. Sekonyela and C. C. C. Wang, Proc. Natl. Acad. Sci. U. S. A., 2013, 110, 17065–17070.
- 50 S. Dhingra, A. L. Lind, H. C. Lin, Y. Tang, A. Rokas and A. M. Calvo, PLoS One, 2013, 8, 1–16.
- 51 N. Grahl, S. Puttikamonkul, J. M. Macdonald, M. P. Gamcsik, L. Y. Ngo, T. M. Hohl and R. A. Cramer, PLoS Pathog., 2011, 7, e1002145.
- 52 T. S. Norton, Q. Al Abdallah, A. M. Hill, R. V. Lovingood and J. R. Fortwendel, Virulence, 2017, 8, 1401–1416.
- 53 M. Kroll, F. Arenzana-Seisdedos, F. Bachelerie, D. Thomas, B. Friguet and M. Conconi, Chem. Biol., 1999, 6, 689–698.
- 54 A. Paugam, C. Creuzet, J. Dupouy-Camet and P. Roisin, Parasitol. Res., 2002, 88, 785–787.
- 55 R. W. Wheatley, K. K. S. Ng and M. Kapoor, Arch. Biochem. Biophys., 2016, 590, 125–137.
- 56 J. Amich, M. Dümig, G. O'Keeffe, J. Binder, S. Doyle, A. Beilhack and S. Krappmann, Infect. Immun., 2016, IAI.01124.
- 57 R. D. Eichner, P. Waring, A. M. Geue, A. W. Braithwaite and A. Mullbacher, J. Biol. Chem., 1988, 263, 3772–3777.
- 58 A. M. Hurne, C. L. L. Chai and P. Waring, J. Biol. Chem., 2000, 275, 25202–25206.
- 59 A. Müller, J. Eller, F. Albrecht, P. Prochnow, K. Kuhlmann, J. E. Bandow, A. J. Slusarenko and L. I. O. Leichert, J. Biol. Chem., 2016, 291, 11477–11490. Netsleaming S7 R. D. Eichner, P. World, A. M. Great, A. W. Franklyncia and 61 S. J. Luidof, D. J. This article. A. Molto Downloaded at the Martin 2019 2:35:2019. Downloaded Universed under Access Articles. The Molton-Non-C
	- 60 E. Cho, J.-J. Hwang, S.-H. Han, S. J. Chung, J.-Y. Koh and J.-Y. Lee, Neurotoxic. Res., 2010, 17, 156–166.
- 61 S. J. Lulloff, B. L. Hahn and P. G. Sohnle, J. Lab. Clin. Med., 2004, 144, 208–214.
- 62 A. Krężel and W. Maret, Arch. Biochem. Biophys., 2016, 611, 3–19.
- 63 A. E. Martell and R. M. Smith, Critical Stability Constants, Plenum Press, New York, 1974.
- 64 P. Laskaris, A. Atrouni, J. A. Calera, C. d'Enfert, H. Munier-Lehmann, J.-M. Cavaillon, J.-P. Latge and O. Ibrahim-Granet, Antimicrob. Agents Chemother., 2016, 60, 5631–5639.